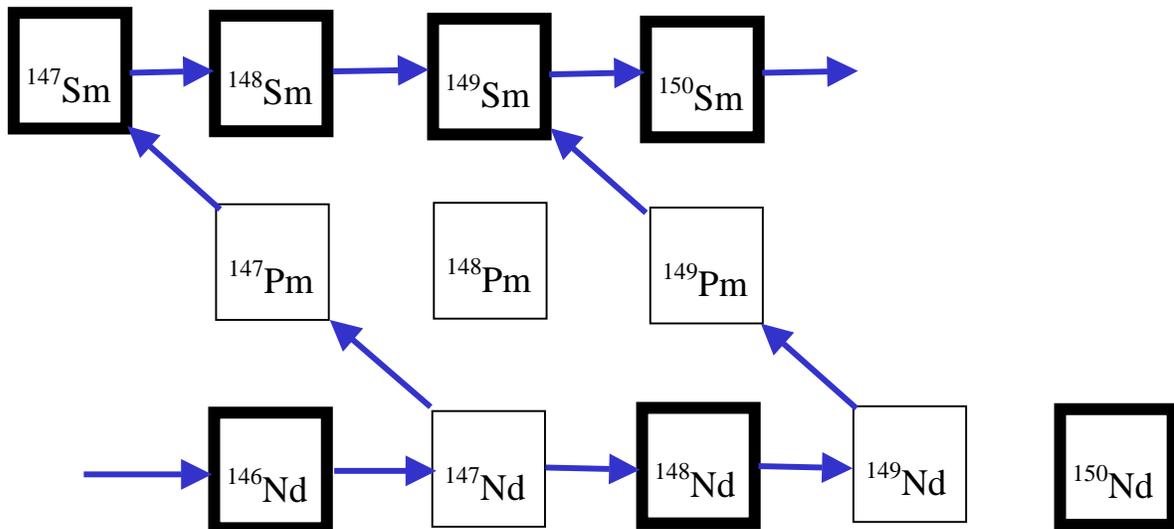


1. The s-process branches at ^{147}Nd because this isotope has a beta decay half life of 5.3 days (under stellar conditions, this is slightly modified compared to what you would find in a table of terrestrial rates), which is not too many orders of magnitude away from the neutron capture timescale. Below is a simplified scheme of the reaction flows in the Nd/Sm region during the s-process (stable isotopes are marked with thick lines)



As you can see ^{148}Sm and ^{150}Sm are s-only isotopes. The neutron capture rates for a stellar neutron spectrum corresponding to a temperature of $kT=30$ keV (typical for s-process) have been measured in the laboratory. The ratio of the ^{148}Sm neutron capture rate to the ^{150}Sm neutron capture rate could be determined particularly accurately and was found to be 0.596. This laboratory result can be used directly to determine the neutron density during the s-process (see Winters et al. *ApJ*300(1986)41 to see how this is really done – here we simplify the problem considerably).

- a. [5 pts] Determine the fraction of the reaction flow that branched into the beta decay at ^{147}Nd during the s-process that produced the solar system abundance distribution. You can use the steady flow approximation (which is locally a very good approximation) that tells you that the reaction flow is proportional to $Y\lambda$, where Y is the observed s-process abundance of the nuclide and λ is the rate of destruction per target nucleus. This means you can measure the (relative) flow that occurred during the s-process at any isotope provided you know Y and λ . Pick the suitable isotopes and work with ratios only.
- b. [5 pts] Using the measured neutron capture rate for ^{147}Nd at a temperature of $kT=30$ keV of $2e7$ cm³/s/mole determine the neutron density during the s-process.